

B12 (2004 G.3)  $z(h) = \frac{1}{N(h)!} \left( \frac{2V}{\lambda^3} e^{-\beta mgh} \right)^{N(h)}$  (spr)

a)  $F(h) = kT N(h) \left[ \ln \frac{N(h) \lambda^3}{2} - 1 \right] + mgh N(h)$

$\mu(h) = \frac{\partial F(h)}{\partial N(h)} = kT \ln \frac{N(h) \lambda^3}{2} + mgh$

$\mu(h) = \mu(h=0) \rightarrow n(h) = n(0) e^{-\beta mgh}$

b)  $n = \frac{2}{V} \approx \frac{1}{P} e^{\beta(E_{\text{kin}} + mgh - \mu)}$

at  $T=0$ ,  $\mu = E_F$  and is  $h$  independent

at  $h=0$ :  $E_F = \frac{k^2}{2m} [3\pi^2 n(0)]^{2/3}$

at  $h>0$ :  $E_F = \frac{p_F^2}{2m} + mgh$  (we "shift" the chem. pot.)

$\frac{p_F^2(h)}{2m} = \frac{k^2}{2m} [3\pi^2 n(h)]^{2/3}$

$E_F - mgh = \frac{p_F^2(h)}{2m}$   
 $(3\pi^2 n(0))^{2/3} \rightarrow \frac{2m^2gh}{k^2} = (3\pi^2 n(h))^{2/3}$   
 $n(h) = \dots$

c)  $n$  non degenerate:  $kT \approx \frac{p_F^2(h)}{2m} = \frac{1}{2m} (3\pi^2 n(h))^{2/3}$   
 $\approx \frac{1}{2m} [3\pi^2 n(0)]^{2/3} - mgh_c$   
 $= E_F - mgh_c$

but since  $kT \gg E_F$

$E_F \approx mgh_c \rightarrow$

$h_c \approx \frac{E_F}{mg}$

from (a)  $\mu = kT \ln(n(h) \lambda^3) + mgh$   
 $h \gg h_c$

at  $h=0$   $\mu = E_F = \frac{1}{2m} [3\pi^2 n(0)]^{2/3} \rightarrow n(h) = \frac{1}{\lambda^3} e^{-\beta mgh + \beta \frac{E_F}{mg}}$