## E3570: Fermi gas in 2D +3 D connected boxes with gravitation

## Submitted by: Avry Shirakov \& Rotem Kupfer

## The problem:

Consider a mesoscopic box that has dimensions $L \times L \times \ell$, such that $\ell \ll L$. In the box there are $N$ spin $1 / 2$ electrons. The mass of an electron is $m$. In items (a-d) assume that the temperature is $T=0$. In items (1-4) the box is attached to a tank that has dimensions $L \times L \times L$, that is placed at height $D$ relative to the box, and you have to take into account the gravitational field $g$. Express your answers using $m, L, \ell, g, T$.
(1) Describe the single particle density of states. Specify the energy range over which it is the same as for a two dimensional box.
(2) Find the fermi energy $E_{F}$ assuming that it is in the range defined above. What is the maximum number $N_{\max }$ of electrons that can be accommodated without violating the $2 D$ description?
(3) Assuming $N<N_{\text {max }}$ find the pressure $P$ on the side walls of the box, and the force $F$ on the horizontal walls.
(4) Assume $N=N_{\max }$. What is the minimum height $D_{\min }$ to place the tank such that all of the electrons stay in the box?
(5) Assume $N=N_{\max }$ and $D>D_{\min }$. The temperature of the system is raised a little bit. As a result some of the particles that were in the $2 D$ box are transferred to the tank. Estimate their number $N^{\prime}$. You are allowed to use any reasonable approximation.

## The solution:

(1) $3 D$ density of states for spin $\frac{1}{2}$ fermions is:

$$
\begin{equation*}
g_{3 D}(\varepsilon)=2 L^{2} l \cdot \frac{(2 m)^{\frac{3}{2}}}{(2 \pi)^{2}} \varepsilon^{\frac{1}{2}} \tag{1}
\end{equation*}
$$

In $2 D$ however, density of states is independent of energy:

$$
\begin{equation*}
g_{2 D}(\varepsilon)=\frac{m L^{2}}{\pi} \tag{2}
\end{equation*}
$$

The condition on the energy for $2 D$ approximation of $3 D$ box is that it is lower then the first excited state of the "short" axis:

$$
\begin{equation*}
\frac{\pi^{2}}{m L^{2}}+\frac{\pi^{2}}{2 m l^{2}} \leq \varepsilon \leq \frac{4 \pi^{2}}{2 m l^{2}}+\frac{\pi^{2}}{m L^{2}} \tag{3}
\end{equation*}
$$

(2) The maximal number of particles in $2 D$ box is given by:

$$
\begin{equation*}
N=\int_{0}^{\infty} g(\varepsilon) f(\varepsilon-\mu) d \varepsilon=\int_{0}^{\infty} \frac{m L^{2}}{\pi} \cdot \frac{1}{\frac{1}{z} e^{\beta \varepsilon}+1} d \varepsilon=\frac{L^{2} m}{\pi} k T \ln \left(1+e^{\beta \mu}\right) \tag{4}
\end{equation*}
$$

For $T \rightarrow 0$ :

$$
\begin{equation*}
\lim _{T \rightarrow 0} k T \ln \left(1+e^{\beta \mu}\right)=\mu \tag{5}
\end{equation*}
$$

So the number of particles at temprature $T=0$ is:

$$
\begin{equation*}
N=\frac{L^{2} m \mu}{\pi} \tag{6}
\end{equation*}
$$

Also at $T=0$ the chemical potential is defined as:

$$
\begin{equation*}
\varepsilon_{F}=\mu \tag{7}
\end{equation*}
$$

So:

$$
\begin{equation*}
\varepsilon_{F}=\frac{\pi N}{m L^{2}} \tag{8}
\end{equation*}
$$

and the maximal number of particles under this approximation is:

$$
\begin{equation*}
N_{\max }=\frac{L^{2} m}{\pi}\left(\frac{4 \pi^{2}}{2 m l^{2}}+\frac{\pi^{2}}{m L^{2}}\right) \approx 2 \pi \frac{L^{2}}{l^{2}} \tag{9}
\end{equation*}
$$

(3) First we notice that at $T=0, F=E$.

The single particle energy function:

$$
\begin{equation*}
\varepsilon_{k_{x}, k_{x}, n_{z}}=k_{x}^{2}+k_{y}^{2}+\frac{1}{2 m}\left(\frac{\pi}{l} n_{z}\right)^{2} \tag{10}
\end{equation*}
$$

We notice that the system is at the ground state for it's "short" axis (it is only dependent on $l$ ) with $n_{z}=1$. The force on the horizontal walls is:

$$
\begin{equation*}
F=-\frac{\partial E}{\partial l}=-N \frac{\partial\left(\frac{\pi^{2}}{2 m l^{2}}\right)}{\partial l}=\frac{N \pi^{2}}{m l^{3}} \tag{11}
\end{equation*}
$$

The energy is given by:

$$
\begin{equation*}
E=\int_{0}^{\infty} g_{2 D}(\varepsilon) \cdot \varepsilon \cdot f(\varepsilon-\mu) d \varepsilon=\int_{0}^{\infty} \frac{m L^{2}}{\pi} \cdot \frac{\varepsilon}{\frac{1}{z} e^{\beta \varepsilon}+1} d \varepsilon \tag{12}
\end{equation*}
$$

By taking $T=0$ and $\mu=\varepsilon_{F}$ we get:

$$
\begin{equation*}
E=\frac{\pi}{2 m} \frac{N^{2}}{L^{2}} \tag{13}
\end{equation*}
$$

and the force on the side walls is:

$$
\begin{equation*}
F=-\frac{\partial E}{\partial L}=-\frac{\partial\left(\frac{\pi}{2 m} \frac{N^{2}}{L^{2}}\right)}{\partial L}=\frac{\pi N^{2}}{m L^{3}} \tag{14}
\end{equation*}
$$

The pressure is given by:

$$
\begin{equation*}
P=\frac{F}{A}=\frac{\pi N^{2}}{A m L^{3}}=\frac{\pi N^{2}}{m L^{4} l} \tag{15}
\end{equation*}
$$

(4) The condition for no transition between the box and the tank is that the chemical potential of the unoccupied tank is higher then the chemical potential of the box:

$$
\begin{align*}
& \mu_{b o x}=\varepsilon_{F}=\frac{\pi N_{\max }}{m L^{2}}  \tag{16}\\
& \mu_{\operatorname{tank}}\left(N^{\prime}=0\right)=m g d  \tag{17}\\
& \mu_{\text {box }}<\mu_{\text {tank }}\left(N^{\prime}=0\right)  \tag{18}\\
& d_{\min }>\frac{\pi N_{\max }}{m^{2} L^{2} g} \tag{19}
\end{align*}
$$

(5) Now we assume $T>0, d>d_{\text {min }}$. The number of particles in the box is given by:

$$
\begin{align*}
& N=\frac{L^{2} m}{\pi} k T \ln \left(1+e^{\beta \mu}\right)  \tag{2}\\
& \mu=k T \ln \left(e^{\frac{\pi N}{L^{2} m k T}}-1\right)=k T \ln \left(e^{\frac{\varepsilon_{f}}{k T}}-1\right)  \tag{21}\\
& \mu=k T \ln \left(e^{\frac{\varepsilon_{f}}{k T}}\left(1-e^{-\frac{\varepsilon_{f}}{k T}}\right)\right)=k T\left(\frac{\varepsilon_{f}}{k T}\right)+k T \ln \left(1-e^{-\frac{\varepsilon_{f}}{k T}}\right) \tag{22}
\end{align*}
$$

By using Taylor expansion near $T=0$, (Note: $e^{-\frac{\varepsilon_{f}}{k T}} \rightarrow 0$; taylor expansion for $\ln (1-x) \sim$ $\left.\ln \left(\frac{1}{1+x}\right) \sim-\ln (1+x) \sim-x\right):$

$$
\begin{equation*}
\mu_{b o x}=\varepsilon_{f}-k T e^{-\frac{\varepsilon_{f}}{k T}} \tag{23}
\end{equation*}
$$

We assume low occupation in the the tank so Boltzmann approximation is valid (and also $N-N^{\prime} \approx$ N)

$$
\begin{equation*}
\mu_{t a n k}=m g d+k T \ln \left(\frac{N^{\prime}}{V} \lambda_{T}^{3}\right) \tag{24}
\end{equation*}
$$

We equate the two potential:

$$
\begin{equation*}
m g d+k T \ln \left(\frac{N^{\prime}}{V} \lambda_{T}^{3}\right)=\varepsilon_{F}-k T e^{-\frac{\varepsilon_{f}}{k T}} \tag{25}
\end{equation*}
$$

In order to get an analytical solution for $N^{\prime}$ we must neglect the exponential term in the chemical potential of the box:

$$
\begin{equation*}
N^{\prime}=\frac{V}{\lambda_{T}^{3}} e^{\frac{\varepsilon_{F}-m_{g} d}{k T}} \tag{26}
\end{equation*}
$$

One can verify that for $T \rightarrow 0, d=d_{\min }$ we get $N^{\prime}=0$.

