

## E307: Magnetization of a ring with electrons

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### The problem:

A ring with radius  $R$  is placed in a uniform magnetic field  $B$ .

There are 5 electrons with mass  $m$ , charge  $e$  and spin  $\frac{1}{2}$ .

Note that two electrons can't be at the same state.

Let  $E$  be the ground state energy

and  $M(B) = -dE/dB$  be the magnetization

$B, R, m, e, g \approx 2.0023$  are given.

If the electrons are spinless the magnetization is a periodic function with period  $B_0$

Later we will assume that  $B = lB_0 + \Delta B$

where  $l$  is an integer and  $0 < \Delta B \ll B_0$ .

- (1) Write an expression for  $B_0$ .
- (2) Find  $M(B)$  for  $B = 0$ .
- (3) Write the addition to  $M(B)$  for small magnetic field.
- (4) Find  $M(B)$  for large  $l$  when  $\Delta B = 0$ .
- (5) Write the addition to  $M(B)$  for large  $l$  when  $\Delta B \neq 0$ .
- (6) Define large  $l$ .

### The solution:

- (1) The Hamiltonian can be written as:

$$H = \frac{1}{2m} \left( p - \frac{eRB}{2} \right)^2$$

The eigenstates of the H are the momentum states  $|k_n\rangle$  where:  $|k_n\rangle = \frac{2\pi}{2\pi R} n$

The eigenvalues are:  $E_n = \frac{1}{2mR^2} \left( n - \frac{eR^2B}{2} \right)^2$

$$M(B) = -\frac{dE}{dB} = -\frac{e}{2m} \left( n - \frac{eR^2B}{2} \right)$$

Now we can see that  $M(B)$  is a periodic function and the period is:

$$B_0 = \frac{2}{eR^2} .$$

- (2) Because of the spin of the electrons, the Hamiltonian include the zeeman term:

$$H = \frac{1}{2m} \left( p - \frac{eRB}{2} \right)^2 - geB \cdot \frac{S}{2m}$$

The Zeeman term split the energy levels so that the energies which has the same  $n$  is not degenerated anymore.

Let us assume that B is parallel to z axis.

When  $B \rightarrow 0$  every two electrons can have the same n, therefore the total energy is:

$$E_{total} = \frac{5e^2R^2\Delta B^2}{8m} - \left(\frac{e}{2m} + \frac{eg}{4m}\right)\Delta B + \frac{3}{2mR^2}$$

$$\text{and } M(B) = \frac{e}{2m}\left(1 - \frac{5\Delta B}{B_0}\right) + \frac{eg}{4m}$$

(3) the addition to  $M(B)$  for small magnetic field is the suseptability:

$$\chi = \frac{dM(B)}{dB} = -\frac{5e}{2mB_0}$$

(4) When l is large enough, all the electrons would have spin  $|\uparrow\rangle$  because of the Zeeman splitting,

Therefore:

$$E_{total} = \frac{1}{2mR^2}\left(\frac{5e^2R^4\Delta B^2}{4} + 10\right) - \frac{5eg\Delta B}{4m}$$

$$\text{and } M(B) = \frac{e}{2m}\left(0 - \frac{5\Delta B}{B_0}\right) + \frac{5eg}{4m}$$

(5) the addition to  $M(B)$  for small magnetic field is the suseptability:

$$\chi = \frac{dM(B)}{dB} = -\frac{5e}{2mB_0}$$

(6) l is large when the difference between the fifth energy level with spin  $|\uparrow\rangle$  to the lowest energy level

with spin  $|\downarrow\rangle$  is smaller than  $\frac{egB}{2m}$ :

$$\frac{2}{mR^2} < \frac{egB}{2m}, \text{ where: } B = lB_0 = \frac{2l}{eR^2}$$

$$\frac{2}{g} < l$$